**CHUKA** 



### **UNIVERSITY**

## UNIVERSITY EXAMINATION

#### **RESIT/SPECIAL EXAMINATION**

#### EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE

**CHEM 315: CHEMICAL APPLICATIONS** 

**STREAMS:** TIME: 2 HOURS

**DAY/DATE: FRIDAY 05/11/2021** 11.30 A.M – 1.30 P.M

### **INSTRUCTIONS:**

Answer question **One** (Compulsory) and any other **Two** questions

## **QUESTION ONE [30 MARKS]**

(a) Determine the point group of each of the following species

(10 marks)

- (i) CHCl<sub>3</sub>
- (ii) CS<sub>2</sub>
- (iii)  $NO_2$  (iv)  $NO_3$
- (v)  $NH_4^+$

(b) On the basis of symmetry, which of the following molecules cannot have a dipole moment:

(6

marks)

- (i) 1,1-Dichloroethene (ii) Benzene (iii) 1,3,5-Trichlorobenzene (iv) 1,3-Difluorobenzene

(c) Set up the multiplication table for the  $C_{3v}$  point group and confirm that the elements form a mathematical group (8 Marks)

(e) Construct a matrix representation for  $C_{2v}$  point group using the s-orbitals of sulphur and the two oxygens of SO<sub>2</sub> (6 marks)

### **QUESTION TWO [20 MARKS]**

(d) Consider the following multiplication table for the  $C_{4v}$  point group:

$C_{4v}$	E	$C_4$	$C_2$	$C_4^{3}$	$\sigma_{\rm v}$	$\sigma_{\rm v}$ '	$\sigma_{d}$	$\sigma_{ ext{d}}'$
Е	Е	$C_4$	$C_2$	$C_4^{3}$	$\sigma_{\rm v}$	$\sigma_{\rm v}$ '	$\sigma_{ ext{d}}$	$\sigma_{ ext{d}}'$
$C_4$	$C_4$	$C_2$	$C_4^{3}$	E	$\sigma_{ m d}$	$\sigma_{d}$	$\sigma_{\rm v}$	$\sigma_{ m v}$
$C_2$	$C_2$	$C_4^{3}$	E	$C_4$	$\sigma_{ m v}$	$\sigma_{\rm v}$	$\sigma_{ m d}$	$\sigma_{ ext{d}}$
$C_4^{3}$	$C_4^3$	E	$C_4$	$C_2$	$\sigma_{\text{d}}$	$\sigma_{\tt d}{}'$	$\sigma_{ m v}$ '	$\sigma_{\rm v}$
$\sigma_{\rm v}$	$\sigma_{\rm v}$	$\sigma_{\mathrm{d}}$	$\sigma_{ m v}$	$\sigma_{ m d}$	E	$C_2$	$C_4$	$C_4^3$
$\sigma_{ m v}$ '	$\sigma_{\rm v}$ '	$\sigma_{ m d}$	$\sigma_{\rm v}$	$\sigma_{d}$	$C_2$	E	$C_4^3$	$C_4$
$\sigma_{\text{d}}$	$\sigma_{\rm d}$	$\sigma_{ m v}$ '	$\sigma_{ m d}$	$\sigma_{\rm v}$	$C_4^{3}$	$C_4$	E	$C_2$
$\sigma_{d}$	$\sigma_{\rm d}$ '	$\sigma_{\rm v}$	$\sigma_{d}$	$\sigma_{\rm v}$	$C_4$	$C_4^{3}$	$C_2$	E

(i) Construct the multiplication table for each of the non-trivial sub-groups of the  $C_{4v}$  point group. (8 marks)

(ii) Determine the classes of the  $C_{4v}$  point group

(6 marks)

(b) Consider the structures of the three isomers of IF<sub>3</sub>O<sub>2</sub>. Determine the symmetry elements and the point group of each isomer (6 marks)

$$F = \bigcup_{i=1}^{N} F_{i} \qquad O = \bigcup_{i=1}^{N} \bigcup_{i=1}^{N} \bigcap_{i=1}^{N} \bigcap_{i=1}^{N} F_{i} \qquad O = \bigcup_{i=1}^{N} \bigcup_{i=1}^{N} \bigcap_{i=1}^{N} \bigcap_{i=1}^{N}$$

# **QUESTION THREE [20 MARKS]**

(a) Determine the fundamental vibrational modes of  $BCl_3(D_{3h})$  that are infrared and Raman active (10 marks)

(b) Determine the atomic orbitals that Xe can use to form hybrid orbitals in  $XeF_4(\mathbf{D}_{4h})$  (10 marks)

### **QUESTION FOUR [20 MARKS]**

(a) Define each of the following terms:

(6 marks)

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(i) symmetry operation	(ii) symmetry element	(iii) rotational axis	
(iv) plane of symmetry	(v) center of symmetry	(vi) improper rotatio	n axis
(b) Discuss the composition of a	character table	(8	8 marks)
(c) Derive the characters of the ir Orthogonality Theorem	reducible representation of th	1 0 1	ng the Great (6 marks)