CHUKA



UNIVERSITY

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EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE IN CHEMISTRY

CHEM 446: CHEMISTRY OF NATURAL PRODUCTS

STREAMS: BSC. CHEMISTRY TIME: 2 HOURS

DAY/DATE: TUESDAY 21/09/2021 11.30 A.M. – 1.30 P.M.

INSTRUCTIONS:

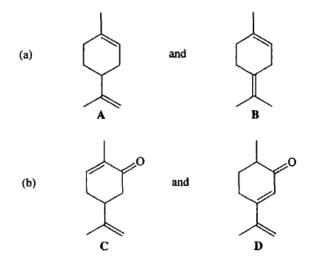
• Answer questions one (compulsory) and any other two questions QUESTION ONE [30 MARKS]

- (a) Explain, with the aid of suitable examples, five commercial uses of alkaloids (5 marks)
- (b) Identify the isoprene unit(s) in the following compounds (6 marks)

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(c) Describe, with the aid of suitable equations, the biosynthesis of borneol from geranyl pyrophosphate (5 marks)

(d) Explain how each of the following pairs can be distinguished using a physical and a chemical method (4 marks)



(e) Discuss the extraction, separation and purification of alkaloids from plant materials

(10

marks)

QUESTION TWO [20 MARKS]

(a) Identify the acetate units in each of the following compounds (3 marks)

(b) Describe, with the aid of suitable equations, the biosynthesis of campherenol from nerolidyl pyrophosphate (4 marks)

(c) Explain, with the aid of suitable examples, six commercial uses of terpenes and terpenoids

(3

marks)

(d) Describe the formation of isopentenyl diphosphate through the mevalonate pathway

(10

marks)

QUESTION THREE [20 MARKS]

- (a) (c) A volatile plant product \mathbf{M} , $C_8H_{14}O$, has strong IR absorption at 1717 cm⁻¹. It possess ¹H NMR signals at δ_H 1.65 (3H, s), 2.15 (3H, s), 2.4 (4H, m) and 5.20 (1H, t). On ozonolysis, compound \mathbf{M} gives, among other products, propanone, and on treatment with iodine and alkali it gives triiodomethane (iodoform). What structural formula can be obtained from:
- (i) Its molecular formula (2 marks)
- (ii) Its IR spectrum (1 marks)
- (iii) Its ¹H NMR spectrum (**3 marks**)
- (iv) Its chemical reactions (2 marks)
- (v) Deduce the structure of compound M (2 marks)

(b) Discuss the extraction and isolation of terpenes and terpenoids from plant materials

(10

marks)

QUESTION FOUR [20 MARKS]

Alkaloid **G**, C₈H₁₃NO, is a tertiary base. It has IR absorption at 3500 cm⁻¹ and ¹H NMR signals at δ 3.0 (5H, m), 3.8 (2H, s) and 5.4 (1H, t, J = 7 Hz). It can be oxidized under mild conditions with MnO₂ to give compound **H**, C₈H₁₁NO, which had an IR absorption at 1680 cm⁻¹ and a UV absorption at 229 nm. Compound **H** formed a deep-red dinitrophenylhydrazone. On catalytic hydrogenation, **G** absorbed 1 mole of hydrogen to give **I**. This readily formed a monotoluene–*p*-sulfonate, which on reduction with LiAlH₄ gave **J**, C₈H₁₅N. When **J** was submitted to three successive Hofmann degradations with a hydrogenation after each stage, the C₈ hydrocarbon 3-methylheptane was obtained. Deduce the structure of **G** (20 marks)

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