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**CHUKA**



**UNIVERSITY**

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**UNIVERSITY EXAMINATIONS**

**EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE IN  
BIOMEDICAL SCIENCE AND TECHNOLOGY**

**BMET 214: BIOCHEMISTRY OF BIOMOLECULES**

**STREAMS: B.SC BIOMED SCI & TECH Y2S1**

**TIME: 2 HOURS**

**DAY/DATE: FRIDAY 13/12/2019**

**8.30 A.M. – 10.30 A.M.**

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**INSTRUCTIONS:**

- **Answer Question One and any other Two Questions**
- **Do not write on the question paper**

**Question One (30 marks)**

- a. Translate the following amino acid sequence into one-letter code: Leu-Glu-Ala-Arg-Asn-Ile-Asn-Gly-Ser-Cys-Ile-Glu-Asn-Cys-Glu-Ile-Ser-Gly-Arg-Glu-Ala-Thr. (2 marks)
- b. Differentiate between the two broad classes of heterocyclic nitrogenous bases that occur in nucleic acids, and provide a structural example of each. (3marks)
- c. Give the structural illustration and indicate whether each of the following sugars consists of anomers, epimers or an aldose – ketose pair. (4 marks)
  - i. D-glyceraldehyde and dihydroxyacetone

- ii. D-glucose and D-mannose
  - iii. D-galactose and D-glucose
  - iv.  $\alpha$ -D-glucose and  $\beta$ -D-glucose
- d. Glucose reacts slowly with hemoglobin and other proteins to form covalent compounds. Explain why glucose is reactive? What is the nature of the adduct formed? (4 marks)
- e. RNA is readily hydrolyzed by alkali, whereas DNA is not. Explain? (5 marks)
- f.  $\alpha$ -D-Mannose is a sweet-tasting sugar.  $\beta$ -D-Mannose, on the other hand, tastes bitter. A pure solution of  $\alpha$ -D-mannose loses its sweet taste with time as it is converted into the  $\beta$  anomer. Draw the  $\beta$  anomer and explain how it is formed from the  $\alpha$  anomer. (6 marks)
- g. Draw the structure of a 16-carbon fatty acid as saturated, trans monounsaturated, and cis monounsaturated and explain why most unsaturated fatty acids found in phospholipids are in the cis rather than trans conformation. (6 marks)

**Question Two (20 marks)**

- g.i.1.a. A mutation that changes an alanine residue in the interior of a protein to valine is found to lead to a loss of activity. However, activity is regained when a second mutation at a different position changes an isoleucine residue to glycine. How might this second mutation lead to a restoration of activity? (6 marks)
- g.i.1.b. All l amino acids have an *S* absolute configuration except l-cysteine, which has the *R* configuration. Explain why l-cysteine is designated as the *R* absolute configuration. (6 marks)
- g.i.1.c. Describe the structure and function of the major proteins in muscle. (8 marks)

**Question Three (20 marks)**

- a. Describe the features of the Watson-Crick model of DNA deduced from the diffraction patterns.

- b. Describe the structural and functional differences between DNA and RNA. (10 marks)
- (10 marks)

**Question Four (20 marks)**

- a. Monosaccharides are reducing sugars. Describe an experiment that can be used to deduce this. (10 marks)
- b. Briefly describe the different classes of lipids. (10 marks)
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