

CHUKA



UNIVERSITY

UNIVERSITY EXAMINATION

FIRST YEAR EXAMINATION FOR THE AWARD OF DEGREE OF BACHELOR OF SCIENCE (GENERAL) & BACHELOR OF EDUCATION (SCIENCE)

PHYS 134: INTRODUCTION TO QUANTUM PHYSICS

STREAMS: BSc. (GEN), B Ed. (SCIENCE) Y1S2

TIME: 2 HOURS

DAY/DATE: FRIDAY 12/04/2019

11.30 A.M – 1.30 P.M

INSTRUCTIONS:

- Attempt question ONE (30 marks) and any other TWO questions (20 marks each).
- Start each question on a fresh page

USEFUL CONSTANTS

Wien's displacement constant $b = 2.898 \times 10^{-3} \text{ m.K}$

Planck's constant $h = 6.63 \times 10^{-34} \text{ J.s} = 4.14 \times 10^{-15} \text{ () eV.s}$

Mass of an electron $m_e = 9.1 \times 10^{-31} \text{ kg}$

Reduced Planck's constant $\hbar = \frac{h}{2\pi} = 1.055 \times 10^{-34} \text{ J.S}$

Electronic charge $e = 1.6 \times 10^{-19} \text{ C}$

QUESTION ONE (Compulsory: 30 Marks)

- a. Using suitable illustrations, explain what a blackbody is in terms of absorption and radiation of heat energy. (4marks)
- b. State Wien's displacement law. (1mark)
- c. Determine the peak wavelength of the blackbody radiation emitted by the Sun, which has a surface temperature of 5800 K. (3marks)
- d. Explain what the half-life of a radioactive substance is. (2marks)

- e. The isotope carbon-14, $^{14}_6\text{C}$, is radioactive and has a half-life of 5 730 years. If you start with a sample of 1 000 carbon-14 nuclei, how many nuclei will still be undecayed in 25000 years? (4 marks)
- f. Estimate the energy of the characteristic x-ray emitted from a tungsten target when an electron drops from an M shell ($n = 3$ state) to a vacancy in the K shell ($n = 1$ state). The atomic number for tungsten is $Z = 74$. (5 marks)
- g. State and explain Heisenberg's uncertainty principle. (3 marks)
- h. The speed of an electron is measured to be 5×10^3 m/s to a precision of 0.003%. Find the minimum uncertainty in determining the position of the electron. (5 marks)
- i. A ball of mass 50g and a de Broglie wavelength of 3.32×10^{-34} m is thrown up. Determine its speed. (3 marks)

QUESTION TWO (Elective: 20 Marks)

- a. State and explain Bohr's postulates that were used to predict the spectrum of the hydrogen atom. (3 marks)
- b. Show that the energy for the quantized states of a hydrogen atom is given by:- (9 marks)

$$E_n = \frac{-13.606}{n^2} e$$

- c. Calculate the wavelength of light for the least energetic photon emitted in the Lyman series of the hydrogen atom spectrum lines. (5 marks)
- d. Determine the series limit for the Lyman series. (3 marks)

QUESTION THREE (Elective: 20 Marks)

- a. State Planck's assumptions in his explanation of blackbody radiation and explain how this explanation relates with experimental observation. (4 marks)
- b. A 2.00-kg block is attached to a massless spring that has a force constant of $k = 25.0$ N/m. The spring is stretched 0.400 m from its equilibrium position and released from rest. Determine:-
- i. the total energy of the system (4marks)
- ii. the frequency of oscillation according to classical calculations. (4 marks)

- c. Assuming the energy of the oscillator is quantized, find the quantum number n for the system oscillating with this amplitude. (4 marks)
- d. Suppose the oscillator makes a transition from the $n = 5.36 \times 10^{33}$ state to the state corresponding to $n = 5.36 \times 10^{33}$, by how much does the energy of the oscillator change in this one-quantum change? (4 marks)

QUESTION FOUR (Elective: 20 Marks)

- a. State and explain the factors affecting the photoelectric effect. (4 marks)
- b. Two light sources are used in a photoelectric experiment to determine the work function for a particular metal surface. When green light incident on the metal surface from a mercury lamp of wavelength 546 nm is used, a stopping potential of 0.4V reduces the photocurrent to zero.
- i. Based on this measurement, what is the work function for this metal? (8 marks)
- ii. What stopping potential would be observed when using the yellow light from a helium discharge tube of wavelength 590 nm? (8 marks)

QUESTION FIVE (Elective: 20 Marks)

- a. Using a suitable illustration, explain what the Compton effect is. (3 marks)
- b. X-rays having energy of 300 keV undergo Compton scattering from a target. The scattered rays are detected at 37° relative to the incident rays. Determine:-
- i. The Compton shift at this angle (5 marks)
- ii. The energy of the scattered X-rays (6 marks)
- iii. The energy of the recoiling electron (6 marks)
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